

Naming*Ternary Ionic Compounds***Learning Target**

Apply the rules for writing formulas of binary ionic compounds.

Practice writing ternary ionic compounds using the table below. Make sure to include the correct charges on each cation and anion.

	Compound	Cation	Anion	Chemical Name
1	K_2CO_3			
2	KOH			
3	$Ca(NO_3)_2$			
4	$SrSO_4$			
5	$CuHCO_3$			
6	Li_3PO_4			
7	$Fe(OH)_2$			
8	Na_2SO_3			
9	$Sn(NO_2)_4$			
10	$Ca(SCN)_2$			
11	$Cu(CH_3COO)_2$			
12	NH_4OH			
13	$Cs_2S_2O_3$			
14	$Fe_2(CO_3)_3$			
15	$CuMnO_4$			
16	$Fe(OH)_3$			
17	Cu_2O			
18	$Pb(CN)_2$			
19	$(NH_4)_2CO_3$			
20	$Rb_2Cr_2O_7$			